

(19) World Intellectual Property
Organization
International Bureau



(43) International Publication Date
22 January 2004 (22.01.2004)

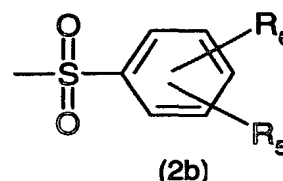
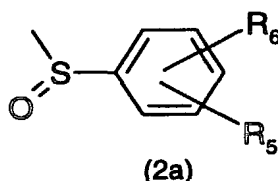
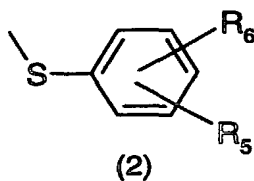
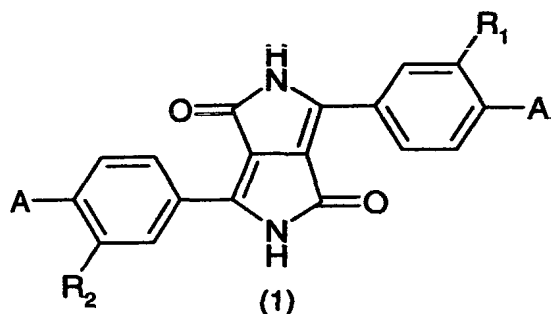
PCT

(10) International Publication Number
WO 2004/007604 A1

- (51) International Patent Classification⁷: **C08K 5/00**, 5/3415, C09B 57/00, 67/00
- (21) International Application Number: PCT/EP2003/007336
- (22) International Filing Date: 8 July 2003 (08.07.2003)
- (25) Filing Language: English
- (26) Publication Language: English
- (30) Priority Data: 1250/02 17 July 2002 (17.07.2002) CH
- (71) Applicant (for all designated States except US): **CIBA SPECIALTY CHEMICALS HOLDING INC.** [CH/CH]; Klybeckstrasse 141, CH-4057 Basel (CH).
- (72) Inventors; and
- (75) Inventors/Applicants (for US only): **LENZ, Roman** [CH/CH]; Tiergartenstrasse 40, CH-4410 Liestal (CH). **DE KEYZER, Gerardus** [NL/CH]; Steingrubenweg 158, CH-4125 Riehen (CH). **WALLQUIST, Olof** [SE/CH]; Spechtstrasse 34, CH-4106 Therwil (CH). **YOUSAF, Taher** [GB/CH]; Efringerstrasse 96, CH-4057 Basel (CH).
- (74) Common Representative: **CIBA SPECIALTY CHEMICALS HOLDING INC.**; Klybeckstrasse 141, CH-4057 Basel (CH).
- (81) Designated States (national): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW.
- (84) Designated States (regional): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).
- Published:
— with international search report

[Continued on next page]

(54) Title: HIGH-MOLECULAR-WEIGHT POLYMERIC MATERIAL COMPRISING DIKETOPYRROLOPYRROLE PIGMENTS



(57) Abstract: The present invention relates to high-molecular-weight polymeric material comprising at least one diketopyrrolopyrrole pigment of formula (1) wherein R_1 , is hydrogen, chlorine, methyl, methoxy, CF_3 or CN, R_2 is hydrogen, chlorine, methyl, methoxy, CF_3 or CN, A is hydrogen, chlorine, methyl, methoxy, CF_3 , CN, unsubstituted or substituted phenyl or a radical of formula (2), (2a) or (2b), wherein R_5 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN and R_6 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN, or R_5 and R_6 together with the phenyl ring to which they are bonded form an aryl or a heteroaryl and A_1 is a radical of formula (2), (2a) or (2b), to the preparation of that material and to the use thereof in colour filters, and to novel diketopyrrolopyrrole pigments.



For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

High-molecular-weight polymeric material comprising diketopyrrolopyrrole pigments

The invention relates to a high-molecular-weight organic material the pigmentation of which consists of extremely fine particles, the pigmentation comprising at least one pigment from the class of diketopyrrolopyrrole compounds, and to novel diketopyrrolopyrrole pigments. That material is preferably employed in the form of thin layers that are built up in one or more layers in the form of patterns on a transparent substrate and can be used as optical colour filters.

Colour filters are an extremely exacting technology which also demands very high standards of the colouring agents. Three pigmentation processes that are fundamentally different in principle are known for the production of colour filters. A number of variants thereof is described in Displays 14/2, 115 (1993).

For example, it is possible to use colorants that can be completely dissolved in a binder, in a conventional solvent or in a solution of the binder in a conventional solvent. According to JP-A-08/44050, a colour strength of ≥ 2.8 is possible for a film thickness of $\leq 1.25 \mu\text{m}$. Disadvantages of the colorants are their inadequate thermal stability and fastness to light.

On the other hand, it is also possible to use dispersions of pigments that are completely insoluble in a binder or in a conventional solvent or are soluble therein only in an extremely low concentration. A number of variants of that method has been disclosed in Journal of the SID 1/3, 341-346 (1993) and at the IDW'95 (Contribution CF-3, pages 57 to 60 of the printed version). In order to obtain the necessary particle fineness corresponding to a very high specific surface, prolonged grinding with the application of high friction energy is required. According to Contribution CF-5 at the IDW'95 (pages 65 to 68 of the printed version), the particle size is to be very fine, $0.1 \mu\text{m}$ maximum.

A third alternative consists of chemically modifying pigments in such a manner that they are completely soluble in a ready-for-use mixture of a binder and a conventional solvent. After coating and the usual further processing, the pigment can be regenerated, the pigment precursors having to undergo a reaction that obviously requires a high activation energy.

The formulation of extremely fine pigments in such a manner that the coloristic properties conforming to exact standards are obtained is therefore extremely difficult by the previously known methods.

- 2 -

The production of colour filters by sequential application of a red, blue and green colouring agent to a suitable substrate, such as, for example, amorphous silicon, is also described in GB-A-2 182 165. The colour filters can be coated, for example, using inks, especially printing inks, comprising the diketopyrrolopyrrole pigments used according to the invention, or can be produced, for example, by mixing the diketopyrrolopyrrole pigments used according to the invention with chemically, thermally or photochemically structurable high-molecular-weight material. The further production can be carried out, for example, analogously to EP-A-654 711 by application to a substrate, e.g. glass used for LCD, subsequent photo-structuring and development. Other documents that describe the production of colour filters are US-A-5 624 467 and WO 98/45756.

For the production of colour filters pigments of the diketopyrrolopyrrole class, so-called DPP pigments, have also been used for some time.

For example, EP-A-704 497 describes mixed crystals and solid solutions of symmetrical diketopyrrolopyrroles that are isomorphous with the crystal structure of an asymmetrical diketopyrrolopyrrole and can be used for pigmenting high-molecular-weight organic substances.

EP-A-654 506 discloses synergistic pigment mixtures that can be used in plastics materials, surface coatings or printing inks, it being possible also for pigments of the diketopyrrolopyrrole class *inter alia* to be used.

EP-A-340 968 discloses an electrolytic process for the production of pigmented thin layers, diketopyrrolopyrrole Red 254 (C.I. Pigment Red 254) *inter alia* being used as pigment.

The alternative method, by which pigments are chemically modified in such a manner that they are completely soluble in a ready-for-use mixture of a binder and a conventional solvent and are regenerated after coating and the usual further processing, is described, for example, in EP-A-654 711 and EP-A-742 255, diketopyrrolopyrrole Red 255 (C.I. Pigment Red 255) *inter alia* being used as pigment.

The DPP pigments used for the production of colour filters up to the present time, however, have the disadvantage *inter alia* that they do not block out a specific blue-tinged red shade with high transparency.

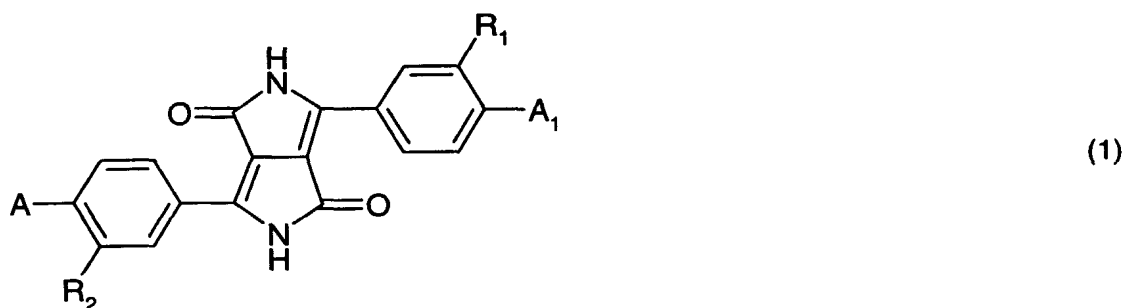
There is therefore a need for new DPP pigments that have such a blue-tinged red shade in

- 3 -

terms of their hue, are coloristically pure and are distinguished by having a high transparency.

Surprisingly, it has now been found that the DPP pigments used according to the invention largely satisfy the above-mentioned criteria.

The present invention accordingly relates to a high-molecular-weight polymeric material comprising at least one diketopyrrolopyrrole pigment (DPP pigment) of formula



wherein

R_1 is hydrogen, chlorine, methyl, methoxy, CF_3 or CN,

R_2 is hydrogen, chlorine, methyl, methoxy, CF_3 or CN,

A is hydrogen, chlorine, methyl, methoxy, CF_3 , CN, unsubstituted or substituted phenyl or a radical of formula



wherein

R_5 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN and

R_6 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN, or

- 4 -

R_5 and R_6 together with the phenyl ring to which they are bonded form an aryl or a heteroaryl and

A_1 is a radical of formula



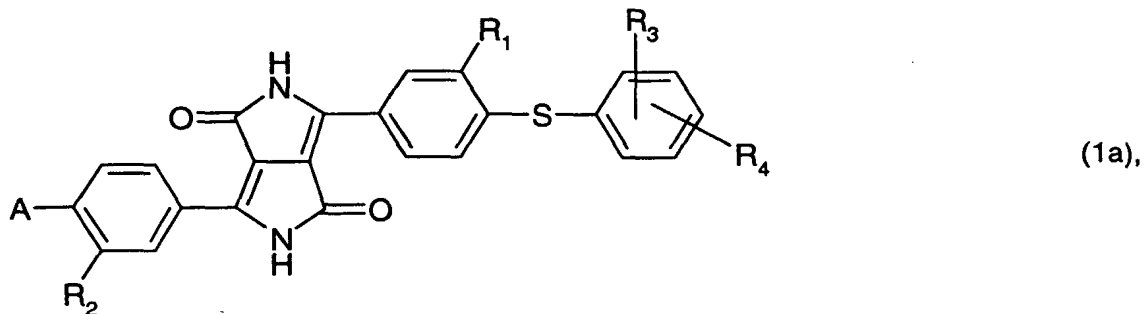
wherein

R_5 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN and

R_6 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN, or

R_5 and R_6 together with the phenyl ring to which they are bonded form an aryl or a heteroaryl.

Preference is given to a high-molecular-weight polymeric material comprising at least one diketopyrrolopyrrole pigment (DPP pigment) of formula



wherein

R_1 is hydrogen, chlorine, methyl, methoxy, CF_3 or CN,

R_2 is hydrogen, chlorine, methyl, methoxy, CF_3 or CN,

R_3 is hydrogen, chlorine, methyl, methoxy and

R_4 is hydrogen, chlorine, methyl, methoxy or

- 5 -

R_3 and R_4 together with the phenyl ring to which they are bonded form a heteroaryl, and A is hydrogen, chlorine, methyl, methoxy, CF_3 , CN, unsubstituted or substituted phenyl or a radical of formula



wherein

R_5 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN and

R_6 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN.

R_1 is preferably hydrogen, chlorine or methyl, especially hydrogen or chlorine.

R_2 is preferably hydrogen, chlorine or methyl, especially hydrogen or chlorine.

R_3 is preferably hydrogen, methyl or methoxy.

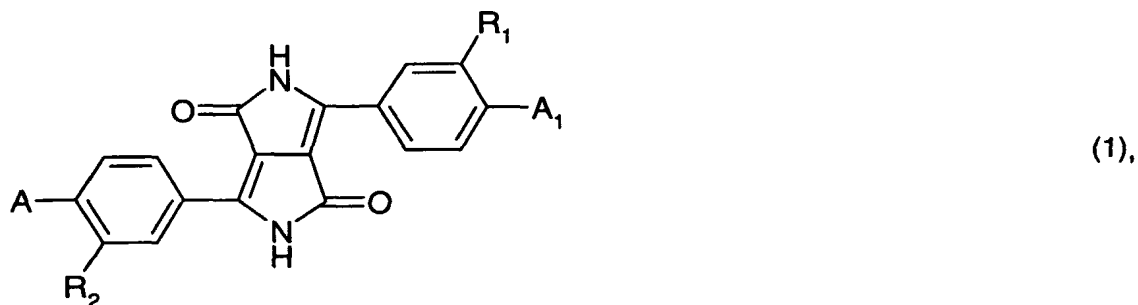
R_4 is preferably hydrogen, methyl or methoxy.

R_5 is preferably hydrogen, methyl or methoxy.

R_6 is preferably hydrogen, methyl or methoxy.

A is preferably hydrogen, chlorine, methyl, phenyl or a radical of formula (2).

The present invention further relates to diketopyrrolopyrrole pigments of formula (1)



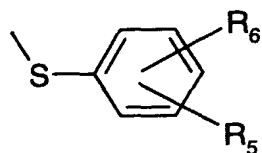
wherein

R_1 is hydrogen, chlorine, methyl, methoxy, CF_3 or CN,

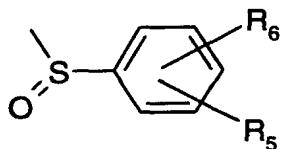
R_2 is hydrogen, chlorine, methyl, methoxy, CF_3 or CN,

A is hydrogen, chlorine, methyl, methoxy, CF_3 , CN, unsubstituted or substituted phenyl or a radical of formula

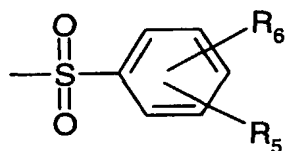
- 6 -



(2),



(2a) or

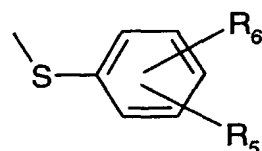


(2b),

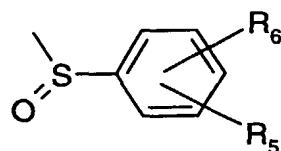
wherein

 R_5 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN and R_6 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN, or R_5 and R_6 together with the phenyl ring to which they are bonded form an aryl or a heteroaryl

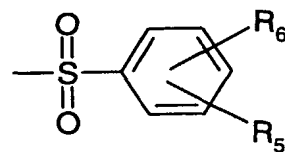
and

 A_1 is a radical of formula

(2),



(2a) or



(2b),

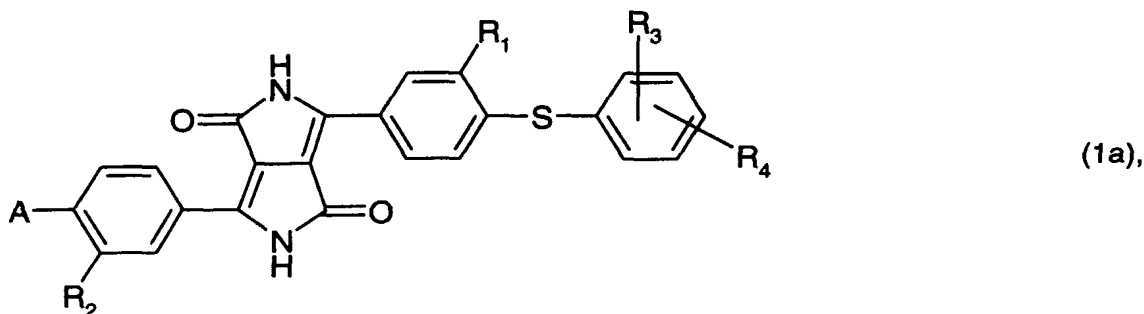
wherein

 R_5 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN and R_6 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN, or

R_5 and R_6 together with the phenyl ring to which they are bonded form an aryl or a heteroaryl, with the proviso that, when both of A and A_1 are a radical of formula (2), R_5 cannot be hydrogen and R_6 cannot be methyl bonded in the 4-position.

- 7 -

Preference is given to diketopyrrolopyrrole pigments of formula



wherein

R₁ is hydrogen, chlorine, methyl, methoxy, CF₃ or CN,

R₂ is hydrogen, chlorine, methyl, methoxy, CF₃ or CN,

R₃ is hydrogen, chlorine, methyl, methoxy and

R₄ is hydrogen, chlorine, methyl, methoxy or

R₃ and R₄ together with the phenyl ring to which they are bonded form a heteroaryl, and

A is hydrogen, chlorine, methyl, methoxy, CF₃, CN, unsubstituted or substituted phenyl or a radical of formula



wherein

R₅ is hydrogen, chlorine, methyl, methoxy, nitro, CF₃ or CN and

R₆ is hydrogen, chlorine, methyl, methoxy, nitro, CF₃ or CN, with the proviso that, when A is a radical of formula (2), R₃ and R₅ cannot be hydrogen and R₄ and R₆ cannot be methyl bonded in the 4-position.

The diketopyrrolopyrrole pigments of formula (1) according to the invention are prepared analogously to generally known preparation methods for diketopyrrolopyrroles as described in US-A-4 579 949, by first reacting, for example, a nitrile of formula



wherein R₁ is as defined above and X is a leaving group, with a compound of formula

- 8 -



or



wherein R_5 and R_6 are as defined above, and then with a succinic acid diester, or by oxidising a compound of formula



resulting from the compounds of formulae (50) and (51) to a compound of formula



or to a compound of formula



and then reacting with a succinic acid diester,
or by first reacting a mixture of two nitriles of formulae



and



wherein R_1 and R_2 are as defined above and X is a leaving group, with a compound of formula

- 9 -



or



wherein R_5 and R_6 are as defined above, and then reacting with a succinic acid diester, or by oxidising a mixture of compounds of formulae



resulting from the compounds of formulae (50), (52) and (51) to a mixture of compounds of formulae



or to a mixture of compounds of formulae



- 10 -



and then reacting with a succinic acid diester.

X as a leaving group is to be understood as being, for example, a fluorine, chlorine, bromine, iodine, nitro or sulfo group.

The two nitriles of formulae (50) and (52) can be used in a molar ratio of from 99:1 to 1:99, and advantageously in a molar ratio of 1:1, to each other.

The nitriles of formulae (50) and (52) and the compounds of formulae (51), (51a), (51b), (51bb), (51c), (51cc), (51d) and (51dd) are known or can be prepared analogously to generally known processes.

The diketopyrrolopyrrole pigments of formula (1) according to the invention might also be prepared analogously to a method described in EP-A-1 073 659.

According to the invention, preference is given to diketopyrrolopyrrole pigments of formula (1) that have a transmission at 570-580 nm of less than 5 % and a transmission at 615 nm of at least 80 % and preferably 85 %.

High-molecular-weight organic materials are understood as being those having a molecular weight of approximately from 10^3 to 10^7 g/mol.

Special preference is given to high-molecular-weight organic material based on acrylates or methacrylates.

The pigmented high-molecular-weight organic material according to the invention is coated onto a transparent substrate as a layer having a thickness of preferably ≤ 20 μm , especially ≤ 10 μm , more especially ≤ 5 μm . In the case of a plurality of layers, each individual layer is preferably ≤ 5 μm in thickness, especially ≤ 2 μm . The substrate preferably has at from 400 to 700 nm a transparency of $\geq 90\%$ throughout, especially $\geq 95\%$, more especially $\geq 98\%$ throughout. Preferably, both the substrate and all layers are flat, the surface roughness thereof being especially ≤ 0.1 μm . Glass or polycarbonate film, for example, can be used as

flat substrates.

Other preferred properties of colour filters are known to the person skilled in the art and are to be found, for example, in the relevant literature or in the publications already mentioned.

Preference is given to a transparent substrate to which a red, a blue and a green layer have been applied in any desired sequence, a red DPP pigment of formula (1) according to the invention being used together with known blue and green compounds.

The treatment that is subsequently carried out where applicable will be one of the conventional, known after-treatments, such as, for example, thermal curing or treatment with a developing bath.

The pigmented high-molecular-weight organic materials according to the invention have a high colour strength, brilliance and transparency. They are eminently suitable for the production of red colour filters for visible light in the range of from 400 to 700 nm.

The invention further relates to a method of modifying the spectrum of a visible light beam by selective absorption of a portion of the radiation by means of a coloured medium, which method comprises using as the coloured medium a pigmented high-molecular-weight polymeric material according to the invention applied to a transparent substrate.

The present invention further relates accordingly to the colour filters produced with the high-molecular-weight polymeric material comprising at least one diketopyrrolopyrrole pigment.

The colour filters according to the invention can be used with excellent results in liquid crystal displays, plasma displays or electro-luminescent displays. The latter may be, for example, active (twistnematic) or passive (supertwistnematic) ferroelectric displays or light-emitting diodes.

The following Examples serve to illustrate the invention. Unless stated otherwise, parts therein are parts by weight and percentages are percentages by weight. Temperatures are given in degrees Celsius. The relationship between parts by weight and parts by volume is the same as that between grams and cubic centimetres:

Example 1a: A mixture consisting of

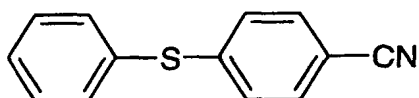
68.8 g (0.5 mol) of 4-chlorobenzonitrile,

103.7 g (0.75 mol) of potassium carbonate and

- 12 -

500 ml of dimethylacetamide

is placed in a laboratory reaction flask. 52.2 g (0.51 mol) of thiophenol are added dropwise to that mixture over a period of 1.5 hours at room temperature and with stirring. The reaction mixture is then stirred at 110°C for 14 hours and is then cooled to room temperature. The cooled reaction mixture is discharged into a two-component mixture consisting of 1000 ml of diethyl ether and 800 ml of water, and the organic phase is separated off and then washed four times with 500 ml of water each time and once with 300 ml of a saturated sodium chloride solution. After drying over magnesium sulfate, the solvent is removed *in vacuo*. 86.4 g (82 % of the theoretical yield) of a compound of formula

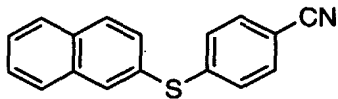
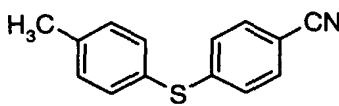
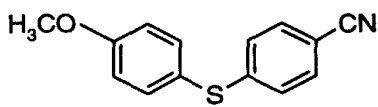
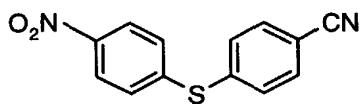
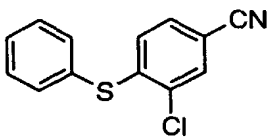
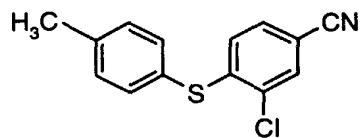


(53)

are obtained.

The compounds listed in Table 1 can be prepared analogously to the process described in Example 1a)

Table 1

starting material 1	starting material 2	compound
2-naphthylthiol	4-chlorobenzonitrile	
4-methylthiophenol	4-chlorobenzonitrile	
4-methoxythiophenol	4-chlorobenzonitrile	
4-nitrothiophenol	4-chlorobenzonitrile	
thiophenol	3,4-dichlorobenzonitrile	
4-methylthiophenol	3,4-dichlorobenzonitrile	

Example 1b: In a laboratory reaction flask,

10.35 g (0.45 mol) of sodium and

100 ml of tert-amyl alcohol are heated under reflux for 12 hours at a temperature of from 110 to 115°C. To the resulting clear solution there is added dropwise at 105°C, with stirring, over a period of 3 hours, a mixture consisting of

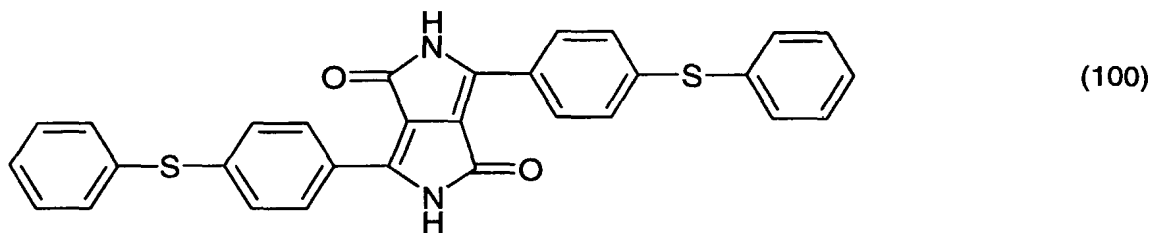
- 14 -

42.3 g (0.2 mol) of the compound of formula (53) from Example 1a,
24.8 ml (0.12 mol) of succinic acid diisopropyl ester and
50 ml of tert-amyl alcohol.

A further 4.14 ml (0.02 mol) of succinic acid diisopropyl ester is then added dropwise over a period of 1 hour. When the addition is complete, the resulting dark-red suspension is maintained at 105°C, with stirring, for a further 30 minutes and is then cooled to 15°C. The suspension is then discharged into a mixture consisting of
500 ml of water,
100 ml of methanol and

36.8 g (0.6 mol) of acetic acid at a temperature below -5°C and is stirred at that temperature for 12 hours to complete the reaction.

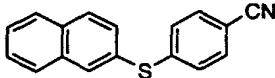
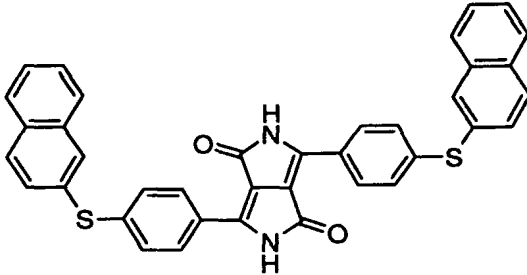
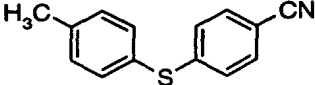
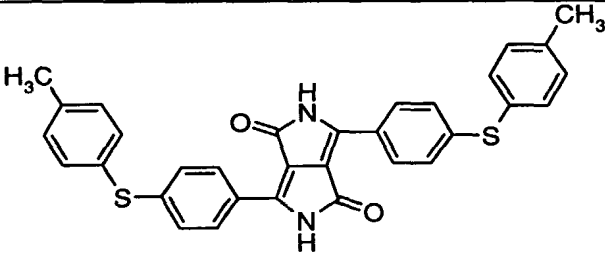
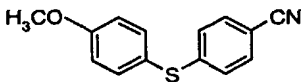
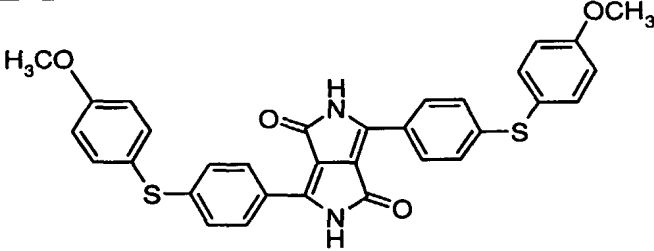
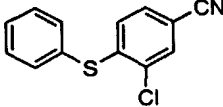
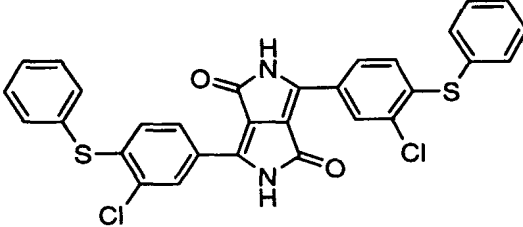
The finely particulate pigment suspension obtained in that manner is filtered off and washed with 1500 ml of methanol and 2000 ml of water. Drying *in vacuo* at 80°C yields 41.1 g (81 % of the theoretical yield) of a red pulverulent compound of formula

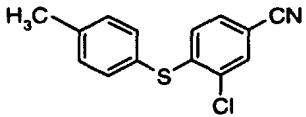
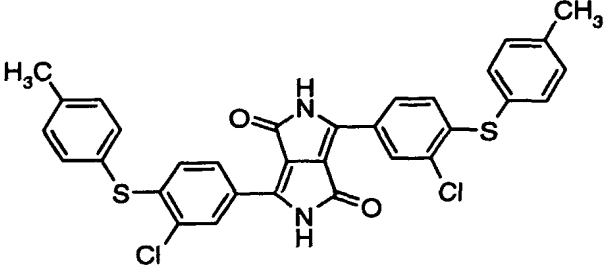


having the spectrum indicated in Fig. 1, which in PVC results in a transparent blue-tinged red coloration.

The diketopyrrolopyrrole pigments listed in Table 2 can be prepared analogously to the process described in Example 1b)

Table 2

starting material	diketopyrrolopyrrole pigment	shade
		blue-tinged red
		blue-tinged red
		blue-tinged red
		blue-tinged red

		blue-tinged red
---	--	--------------------

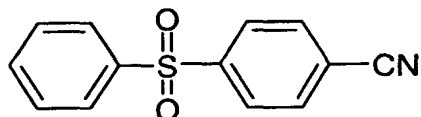
Example 2: A mixture consisting of

34.38 g (0.25 mol) of 4-chlorobenzonitrile,

51.22 g (0.31 mol) of anhydrous sodium benzenesulfinate and

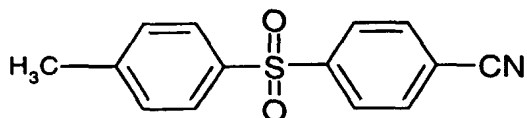
3500 ml of N,N-dimethylacetamide

is placed in a laboratory reaction flask and heated at a temperature of from 140 to 152°C for 5 hours and stirred at 120°C for 15 hours to complete the reaction. The resulting yellow suspension is cooled and the solvent is removed *in vacuo* in a rotary evaporator. The resulting solid residue is taken up in 1.5 litres of water, filtered and washed with 1.4 litres of water. Drying *in vacuo* yields 57.9 g (95 % of the theoretical yield) of a pale-yellow compound of formula



(54).

Example 3: The procedure is analogous to Example 2 except that 56.8 g of anhydrous sodium 4-toluenesulfinate are used instead of the sodium benzenesulfinate. 59.2 g (92 % of the theoretical yield) of a pale-yellow compound of formula



(55)

are obtained.

Example 4: In a laboratory reaction flask,

2.58 g (0.112 mol) of sodium and

60 ml of tert-amyl alcohol

- 17 -

are heated under reflux for 12 hours at a temperature of from 110 to 115°C. To the resulting clear solution there is added dropwise at 105°C, with stirring, over a period of 3 hours a mixture consisting of

12.15 g (0.05 mol) of the compound of formula (54) from Example 2,

9.04 g (0.07 mol) of succinic acid diisopropyl ester and

60 ml of tert-amyl alcohol.

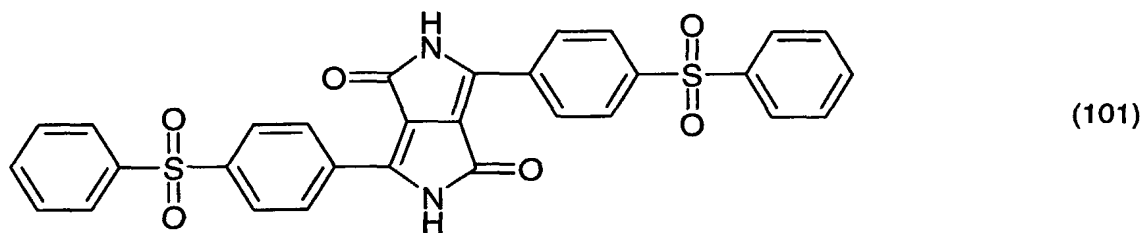
When the addition is complete, the resulting dark-red suspension is maintained at 105°C, with stirring, for a further 30 minutes and is then cooled to room temperature. The suspension is then discharged into a mixture consisting of

250 ml of water,

50 ml of methanol and

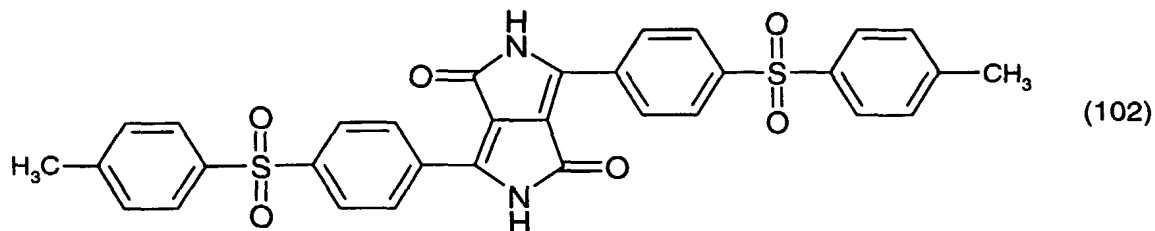
9.0 g (0.15 mol) of acetic acid at a temperature below 30°C and is stirred at room temperature for 15 hours to complete the reaction.

The finely particulate pigment suspension obtained in that manner is filtered off and washed with 1000 ml of methanol and 500 ml of water. After drying *in vacuo* at 90°C, 12.9 g (91 % of the theoretical yield) of a red pulverulent compound of formula



are obtained, which in PVC results in a transparent blue-tinged red coloration.

Example 5: The procedure is analogous to Example 4 except that, instead of compound (54), an equimolar amount of the compound of formula (55) is used. A red pulverulent compound of formula



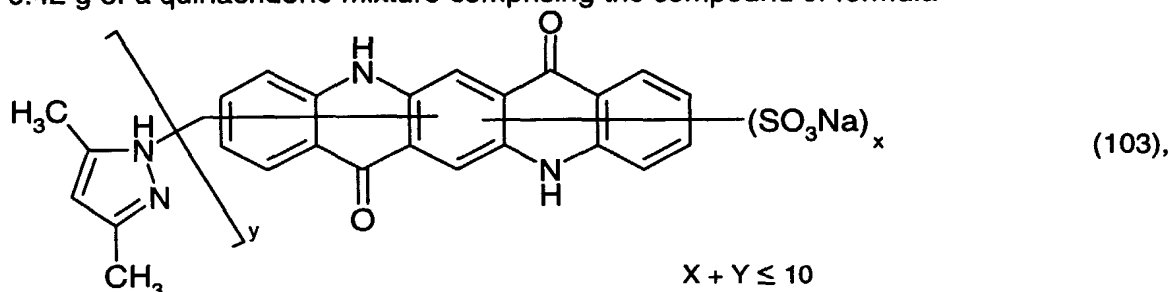
is obtained in a yield of 95 %, which in PVC results in a transparent blue-tinged red coloration.

tion.

Example 6: Production of a red colour filter

8.35 g of the DPP pigment from Example 1b,

0.42 g of a quinacridone mixture comprising the compound of formula



0.42 g of a sulfonated diketopyrrolopyrrole prepared according to Example 1a+b of GB-A-2 238 550,

0.35 g of Solsperse® 22'000 (Avecia),

1.75 g of Solsperse® 24'000 (Avecia) and

40.0 g of propylene glycol monomethyl ether acetate (MPA, CAS Reg. N° 108-65-6)

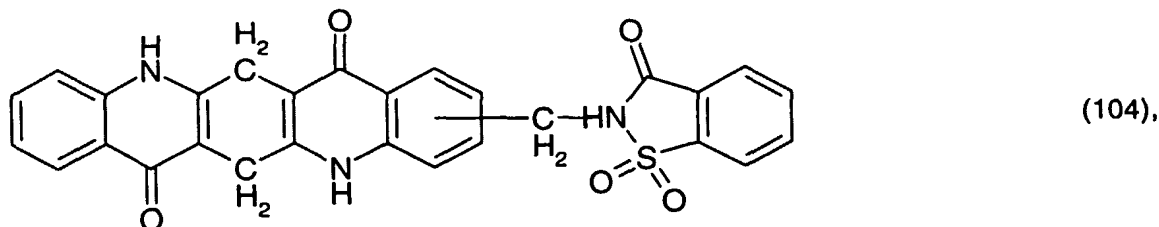
are loaded into a 180 ml glass vessel containing 235 g of zircon ceramic balls and stirred at 20°C, using a Dispermat, at 1000 rev/min for 10 minutes and at 3000 rev/min for 180 minutes. After the addition of 14.3 g of an acrylate polymer binder (40 % solution in MPA), the mixture is further stirred at 3000 rev/min for 30 minutes at room temperature. After removal of the balls, the dispersion is diluted with the same amount by weight of MPA.

On a spin-coating apparatus, a glass substrate (Corning Type 1737-F) is coated with that dispersion and spun at 1000 rev/min for 30 seconds. Drying of the layer is carried out at 100°C for 2 minutes and at 200°C for 5 minutes on a heating plate. The layer thickness obtained is between 0.4 and 0.5 µm.

Example 7: Production of a red colour filter

8.35 g of the DPP pigment from Example 1b,

0.42 g of a dihydroquinacridone compound of formula



- 19 -

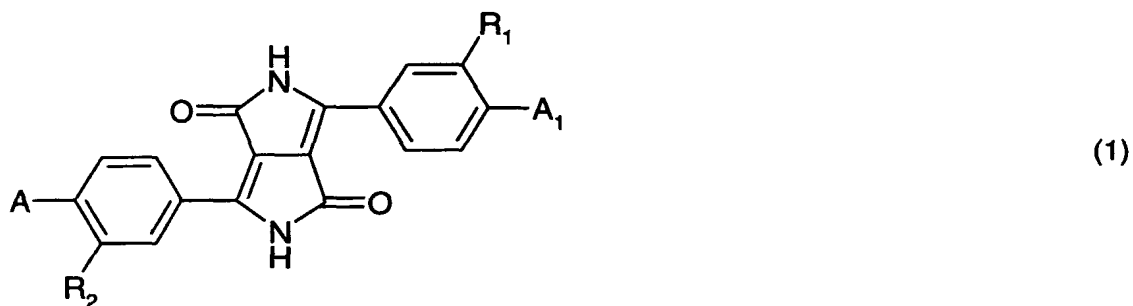
0.42 g of a sulfonated diketopyrrolopyrrole prepared according to Example 1a+b of GB-A-2 238 550,
0.35 g of Solsperse® 22'000 (Avecia),
1.75 g of Solsperse® 24'000 (Avecia) and
40.0 g of propylene glycol monomethyl ether acetate (MPA, CAS Reg. N° 108-65-6)
are loaded into a 180 ml glass vessel containing 235 g of zircon ceramic balls and stirred at 20°C, using a Dispermat, at 1000 rev/min for 10 minutes and at 3000 rev/min for 180 minutes. After the addition of 14.3 g of an acrylate polymer binder (40 % solution in MPA), the mixture is further stirred at 3000 rev/min for 30 minutes at room temperature. After removal of the balls, the dispersion is diluted with the same amount by weight of MPA.

On a spin-coating apparatus, a glass substrate (Corning Type 1737-F) is coated with that dispersion and spun at 1000 rev/min for 30 seconds. Drying of the layer is carried out at 100°C for 2 minutes and at 200°C for 5 minutes on a heating plate. The layer thickness obtained is between 0.4 and 0.5 µm.

The colour filters produced with the diketopyrrolopyrrole pigments used according to the invention are distinguished by a high transmission of the colour dots.

What is claimed is:

1. A high-molecular-weight polymeric material comprising at least one diketopyrrolopyrrole pigment (DPP pigment) of formula



wherein

R_1 is hydrogen, chlorine, methyl, methoxy, CF_3 or CN, R_2 is hydrogen, chlorine, methyl, methoxy, CF_3 or CN, A is hydrogen, chlorine, methyl, methoxy, CF_3 , CN, unsubstituted or substituted phenyl or a radical of formula



wherein

R_5 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN and R_6 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN, or R_5 and R_6 together with the phenyl ring to which they are bonded form an aryl or a heteroaryl and A_1 is a radical of formula

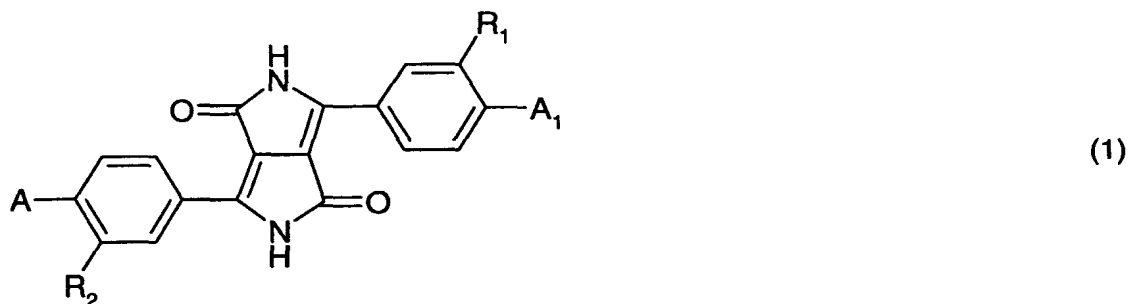
- 21 -



wherein

R_5 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN and R_6 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN, or R_5 and R_6 together with the phenyl ring to which they are bonded form an aryl or a heteroaryl.

2. A diketopyrrolopyrrole pigment of formula

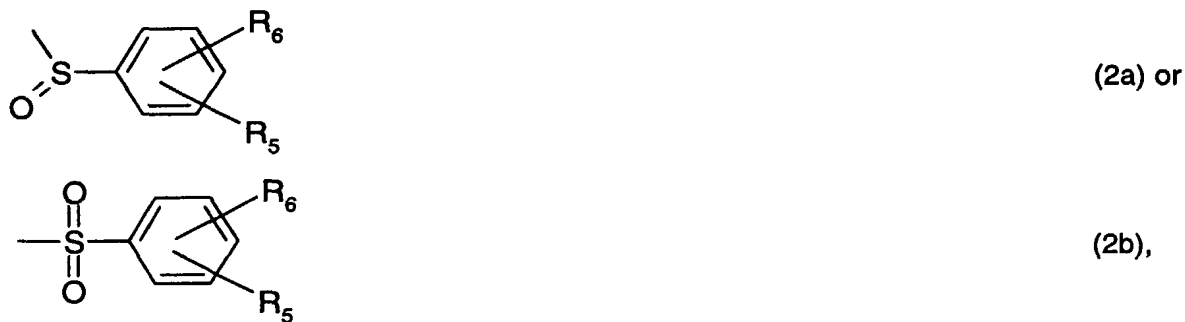


wherein

R_1 is hydrogen, chlorine, methyl, methoxy, CF_3 or CN, R_2 is hydrogen, chlorine, methyl, methoxy, CF_3 or CN, A is hydrogen, chlorine, methyl, methoxy, CF_3 , CN, unsubstituted or substituted phenyl or a radical of formula



- 22 -



wherein

R_5 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN and R_6 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN, or R_5 and R_6 together with the phenyl ring to which they are bonded form an aryl or a heteroaryl and A_1 is a radical of formula

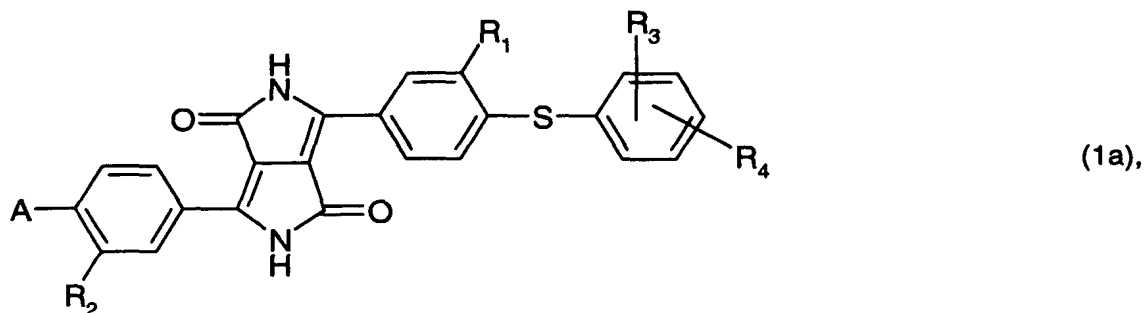


wherein

R_5 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN and R_6 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN, or R_5 and R_6 together with the phenyl ring to which they are bonded form an aryl or a heteroaryl, with the proviso that, when both of A and A_1 are a radical of formula (2), R_5 cannot be hydrogen and R_6 cannot be methyl bonded in the 4-position.

3. A diketopyrrolopyrrole pigment according to claim 2 of formula

- 23 -



wherein

R_1 is hydrogen, chlorine, methyl, methoxy, CF_3 or CN, R_2 is hydrogen, chlorine, methyl, methoxy, CF_3 or CN, R_3 is hydrogen, chlorine, methyl, methoxy and R_4 is hydrogen, chlorine, methyl, methoxy or R_3 and R_4 together with the phenyl ring to which they are bonded form a heteroaryl, and A is hydrogen, chlorine, methyl, methoxy, CF_3 , CN, unsubstituted or substituted phenyl or a radical of formula



wherein

R_5 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN and R_6 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN, with the proviso that, when A is a radical of formula (2), R_3 and R_5 cannot be hydrogen and R_4 and R_6 cannot be methyl bonded in the 4-position.

4. A process for the preparation of a diketopyrrolopyrrole pigment of formula (1) according to claim 2, which comprises first reacting a nitrile of formula



wherein R_1 is as defined above and X is a leaving group, with a compound of formula



or

- 24 -



wherein R_5 and R_6 are as defined above, and then with a succinic acid diester, or oxidising a compound of formula



resulting from the compounds of formulae (50) and (51) to a compound of formula



or to a compound of formula



and then reacting with a succinic acid diester,
or first reacting a mixture of two nitriles of formulae



and



wherein R_1 and R_2 are as defined above and X is a leaving group, with a compound of formula



or

- 25 -



wherein R_5 and R_6 are as defined above, and then reacting with a succinic acid diester, or oxidising a mixture of compounds of formulae



resulting from the compounds of formulae (50), (52) and (51) to a mixture of compounds of formulae



or to a mixture of compounds of formulae

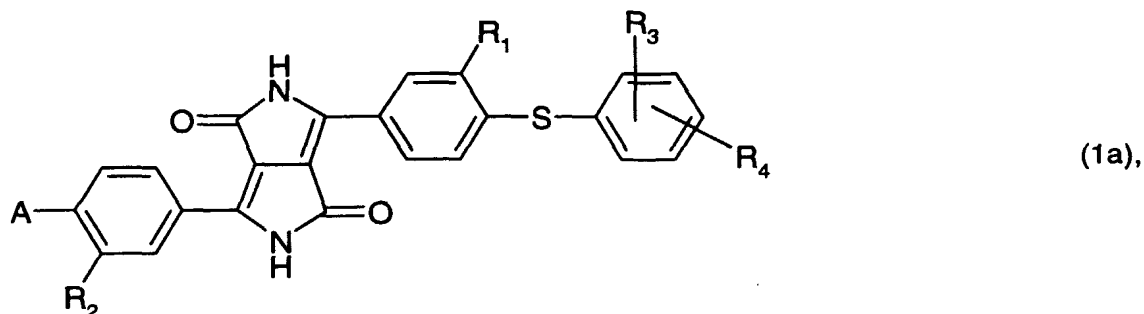


and then reacting with a succinic acid diester.

5. A high-molecular-weight polymeric material comprising at least one diketopyrrolopyrrole

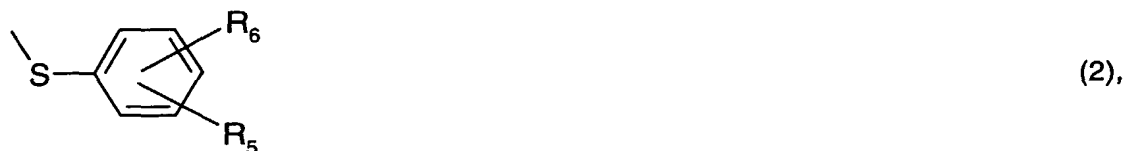
- 26 -

pigment according to claim 3 of formula



wherein

R_1 is hydrogen, chlorine, methyl, methoxy, CF_3 or CN, R_2 is hydrogen, chlorine, methyl, methoxy, CF_3 or CN, R_3 is hydrogen, chlorine, methyl, methoxy and R_4 is hydrogen, chlorine, methyl, methoxy or R_3 and R_4 together with the phenyl ring to which they are bonded form a heteroaryl, and A is hydrogen, chlorine, methyl, methoxy, CF_3 , CN, unsubstituted or substituted phenyl or a radical of formula



wherein

R_5 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN and R_6 is hydrogen, chlorine, methyl, methoxy, nitro, CF_3 or CN.

6. A high-molecular-weight polymeric material according to claim 5, wherein, in formula (1a), R_1 is hydrogen, chlorine or methyl, R_2 is hydrogen, chlorine or methyl, R_3 is hydrogen, chlorine or methyl, R_4 is hydrogen, chlorine or methyl and A is hydrogen, chlorine, methyl or phenyl.

7. A high-molecular-weight polymeric material according to either claim 5 or claim 6, wherein, in formula (1a), A is a radical of formula (2) in which R_5 is hydrogen, methyl or methoxy and R_6 is hydrogen, methyl or methoxy.

8. A high-molecular-weight polymeric material according to claim 1, wherein the high-molecular-weight organic material is based on acrylates or methacrylates.

9. A process for the production of colour filters, wherein a diketopyrrolopyrrole pigment of

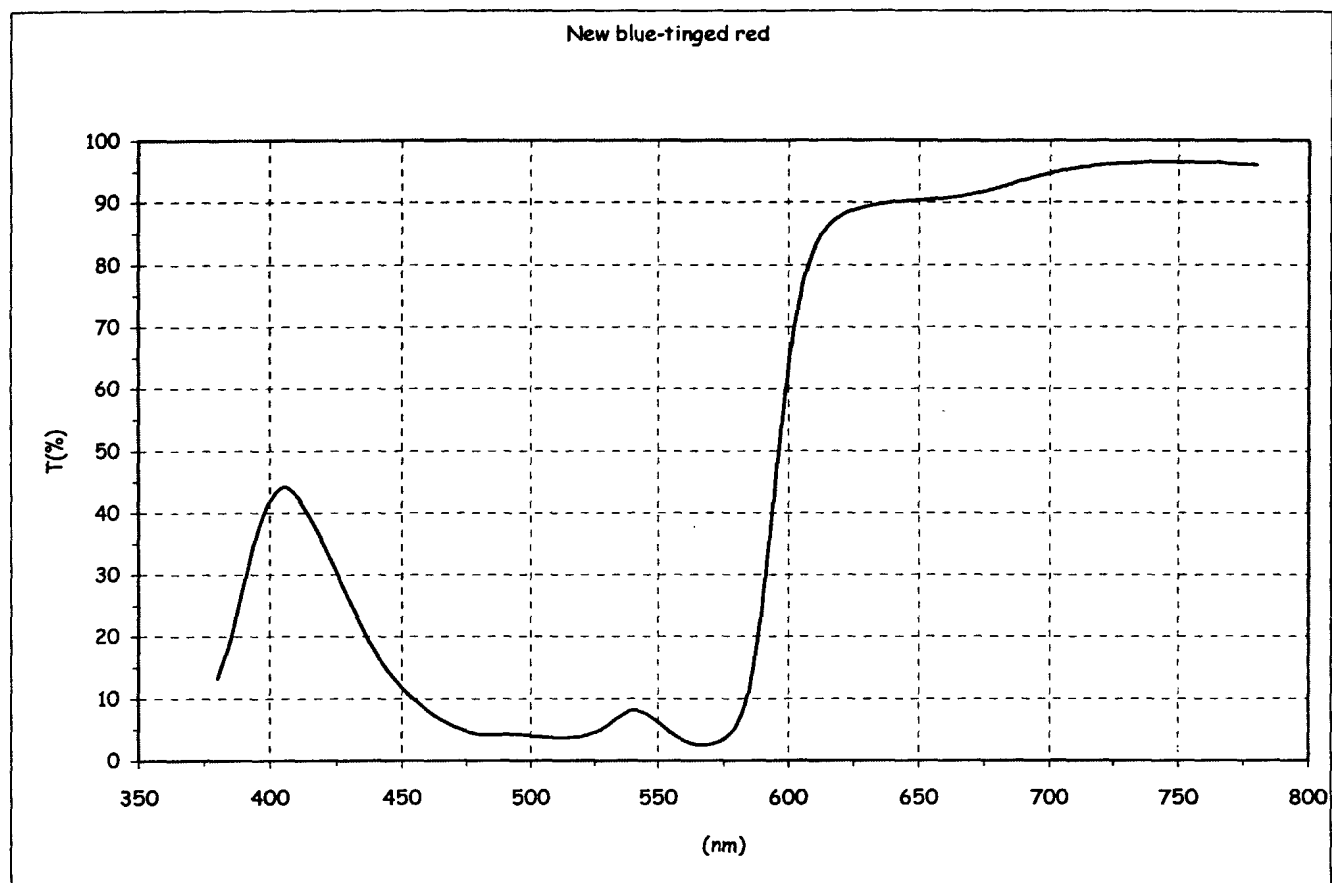
formula (1) according to claim 1 is used.

10. A process for the production of colour filters, wherein a high-molecular-weight polymeric material according to claim 8 is used.

11. The use of a diketopyrrolopyrrole pigment of formula (1) according to claim 1 for the production of colour filters.

12. A colour filter produced with a diketopyrrolopyrrole pigment of formula (1) according to claim 2 or with a high-molecular-weight polymeric material according to claim 1.

I/I

Fig. 1

INTERNATIONAL SEARCH REPORT

International Application No

PCT/EP 03/07336

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C08K5/00 C08K5/3415 C09B57/00 C09B67/00

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C08K C09B

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

CHEM ABS Data, BEILSTEIN Data, EPO-Internal, PAJ, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 0 748 851 A (CIBA GEIGY AG) 18 December 1996 (1996-12-18) abstract; claims 1,5,12	1-3,5-8
Y	page 6, line 7 - line 14 ---	9-12
X	US 4 579 949 A (IQBAL ABUL ET AL) 1 April 1986 (1986-04-01) cited in the application	1-3,5-8
Y	abstract; claims 1-4,7,8 column 4, line 8 - line 28 column 7, line 60 -column 8, line 4 ---	4
Y	WO 02 10288 A (CIBA SC HOLDING AG) 7 February 2002 (2002-02-07) claim 12 page 4, paragraph 3 page 6, paragraph 2 page 9, last paragraph ---	9-12
	-/--	

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

* Special categories of cited documents :

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the international filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

- *T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- *X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- *Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- *&* document member of the same patent family

Date of the actual completion of the international search

29 October 2003

Date of mailing of the international search report

05/11/2003

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl.
Fax: (+31-70) 340-3016

Authorized officer

Schütte, M .

INTERNATIONAL SEARCH REPORT

International Application No.

PCT/EP 03/07336

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	<p>KRICHELDORF H R ET AL: "NEW POLYMER SYNTHESES, 65. \NEW TEREPHTALIC ACIDS WITH AROMATIC SUBSTITUENTS AND THERMOTROPIC POLYESTERS DERIVED FROM THEM" MAKROMOLEKULARE CHEMIE, MACROMOLECULAR CHEMISTRY AND PHYSICS, HUTHIG UND WEPF VERLAG, BASEL, CH, vol. 194, no. 8, 1 August 1993 (1993-08-01), pages 2183-2198, XP000384114 ISSN: 0025-116X formula 16 page 2185, paragraph 3 -----</p>	4

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/EP 03/07336

Patent document cited in search report		Publication date		Patent family member(s)	Publication date
EP 0748851	A	18-12-1996	CA	2179106 A1	16-12-1996
			CN	1146469 A , B	02-04-1997
			CZ	9601736 A3	15-01-1997
			DE	59606863 D1	13-06-2001
			EP	0748851 A2	18-12-1996
			JP	9025428 A	28-01-1997
			US	5738719 A	14-04-1998

US 4579949	A	01-04-1986	AT	22104 T	15-09-1986
			AU	568298 B2	24-12-1987
			AU	1444783 A	24-11-1983
			BR	8302570 A	17-01-1984
			CA	1236105 A1	03-05-1988
			CS	236794 B2	15-05-1985
			DD	209832 A5	23-05-1984
			DD	209832 C4	18-12-1985
			DE	3366011 D1	16-10-1986
			DK	217683 A , B,	18-11-1983
			EP	0094911 A2	23-11-1983
			ES	8406482 A1	01-11-1984
			HU	190489 B	29-09-1986
			JP	1746176 C	25-03-1993
			JP	4025273 B	30-04-1992
			JP	58210084 A	07-12-1983
			KR	9006751 B1	20-09-1990
			PL	242009 A1	13-08-1984
			SU	1225489 A3	15-04-1986
			ZA	8303468 A	29-02-1984

WO 0210288	A	07-02-2002	AU	8974101 A	13-02-2002
			WO	0210288 A1	07-02-2002
			EP	1305371 A1	02-05-2003
			US	2002016999 A1	14-02-2002